



DYS 2002/C002

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Serial No: 10/627,827
Filed: July 24, 2003
Title: NON-AZO DISPERSE DYE MIXTURES

Art Unit: 1712
Examiner: Einsmann

Hon. Commissioner of Patents & Trademarks
Washington, D. C. 20 231

DECLARATION (Rule 132)

Sir:

I, Ono, Shinsuke from Eppstein, declare: I am a chemical engineer, and a citizen of Japan, residing at Waldallee 7 Appartement 2, Eppstein, Federal Republic of Germany.

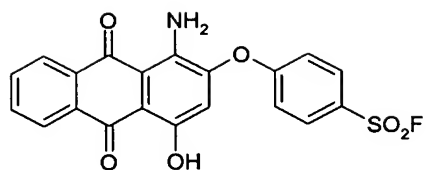
Since completing my studies at Kanazawa University in Japan and having taken my bachelor's degree in 2003, I have been employed as a dyeing application engineer by Hoechst Japan Limited, DyStar Japan Limited afterwards and DyStar Textilfarben GmbH & Co. Deutschland KG at the present. I have had adequate professional experience in the field to which patent application Serial No. 10/627,827, filed July 24, 2003, pertains and which was filed by David Brierley, Alan Leaver, Nigel Hall and Alan Cunningham.

I further declare:

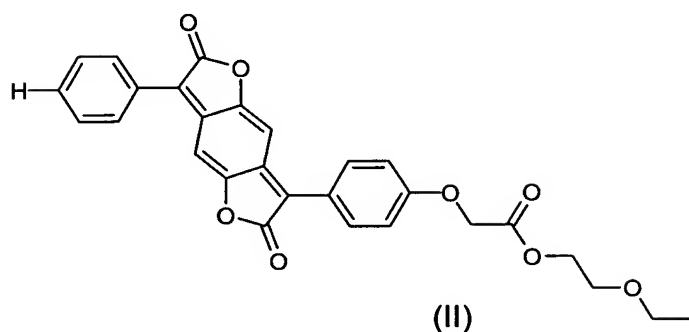
In order to demonstrate that the dye mixtures according to the present application are not obvious over the teachings of WO 95/21958 in view of WO 97/04031 the tests described below have been carried out under my personal guidance and supervision.

I. DYESTUFFS AND DYESTUFF MIXTURES

A.dyestuff according to formula (1) in WO 95/32958



B: dyestuff according to



C: Inventive mixtures of dyestuff A and B

II. PRODUCING OF THE DYEINGS

Dyeings of each of the dyestuffs A, B, and mixture C were produced using a conventional exhaustion process at 120°C for 60 minutes and at 130°C for 60 minutes on PES T5085 (TORAY) 5.00g.

Dyeing conditions:

Liquor ratio: 1:12

pH 4,5 (acetic acid /sodium acetate)

Levelling agent: 1-0% by weight. Levegal DLP (BASF)

Dispersing agent: 1.0% by weight Avolan IS (BASF)

Aftertreatment: Reduction clear

III. Determination of the Temperature dependent build up behaviour

The color depths of the dyeings obtained according to II above were determined by measuring their remission in the visible range of from 400 to 700 nm and determining their overall remission, INTEG value, which is based on K/S value (Kubelka-Munk function) and described below.

$$\text{INTEG value} = \sum_{\lambda=400}^{\lambda=700} I_{\lambda} \cdot K/S_{\lambda} \cdot (x_{\lambda} + y_{\lambda} + z_{\lambda})$$

I_{λ} : the spectral energy distribution of Illuminant

x_{λ} , y_{λ} , z_{λ} : the color-matching functions

The results are given in table1

Dyestuff/mixture	INTEG value 60' at 120°C	INTEG value 60' at 130°C	INTEG value ratio in %
A 0,9% of weight	13,50	20,45	66
C (0,675% A and 0,07% B)	17,39	20,45	85
C (0,370% A and 0,16% B)	19,01	19,79	96
C (0,225% A and 0,21% B)	18,62	20,03	93
B 0,28 of weight	15,44	19,76	78

This improvement of the dye mixture of U.S. Patent Application Serial No: 10/627,827 over the single components could not at all be foreseen and was thus unexpected and surprising. No hint can be found in the prior art for any improvement for this improved build up behavior of the inventive mixture.

I further declare that I understand the contents of this Declaration, that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Signed at Frankfurt

This 04/29 day of 2005

A handwritten signature in black ink, appearing to read 'S. Ono', is written above a horizontal dotted line.

(Shinsuke Ono)